



### LIFE CYCLE ASSESMENT OF TRITIUM IN THE SYSTEM OF THE GRAND CANAL D'ALSACE AND RHINE RIVER

Internship period: 18 February 2019 -18 July 2019

**Ari SAPUTRA** 

Academic Tutor: Olivier PERON Company Supervisor: Mireille DEL NERO

**Gaetana QUARANTA** 

#### INSTITUTE PROFILE

cnrs Institute Université Pluridisciplinaire **Hubert Curien (IPHC)** de Strasbourg Department of Departement of Department of Ecology, Department of Radiobiology, Hadron **Analytical Sciences** Subatomic Research Physiology and Therapy and Molecular Ethology (DEPE) (DRS) (DSA) Imaging (DRHIM)

#### Research Field:

- Nuclear physics and particle physics.
- Radiochemistry.

Radiation protection.

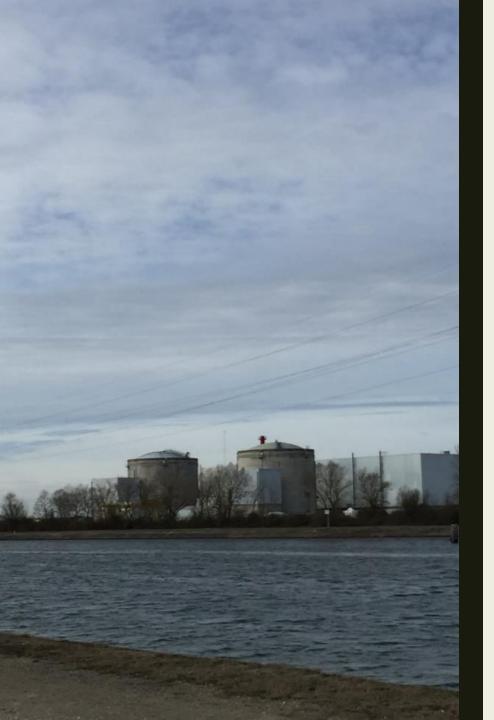
Internship Research Group

#### INTERNSHIP MAIN OBJECTIVE

To calculate a fate factor of Tritium in the system of Rhine river and Grand Canal Alsaces in order to model transfer mechanisms between water and sediments.

#### INTERNSHIP SPECIFIC OBJECTIVES

- Analysis of waters in the Rhine and Grand Canal Alsace systems.
- Analysis of nanoparticles
- Experiments on sediment collected in the Rhine river



#### **SUMMARY**

**INTRODUCTION** 

MATERIAL AND METHODS

WATER CHEMISTRY

**NANOPARTICLES** 

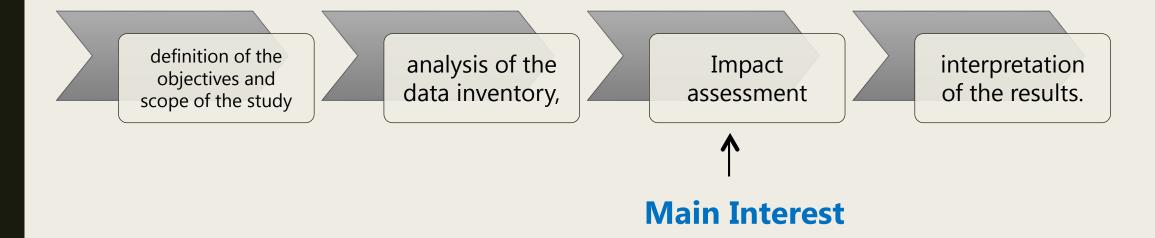
SEDIMENT ADSORPTION EXPERIMENT

**CONCLUSION AND RECOMMENDATION** 

## INTRODUCTION

#### LIFE CYCLE ASSESMENT

■ Life Cycle Assesment (LCA) is a standarized tool for assessing the environmental impacts of elements in a system.



#### LIFE CYCLE ASSESMENT

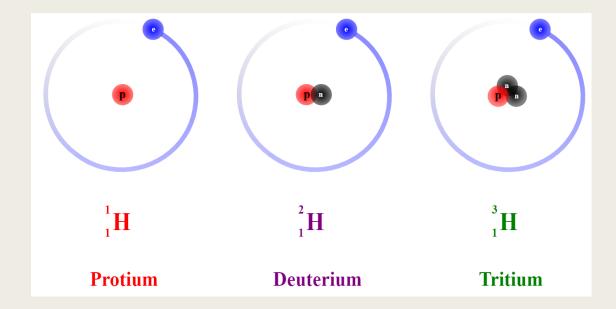
To calculate the pollutant transfer in the environment, it can be charaterize by using mathematical equation:

$$CF = FF \times EF$$

characterization Fate factor Effect Factor factor

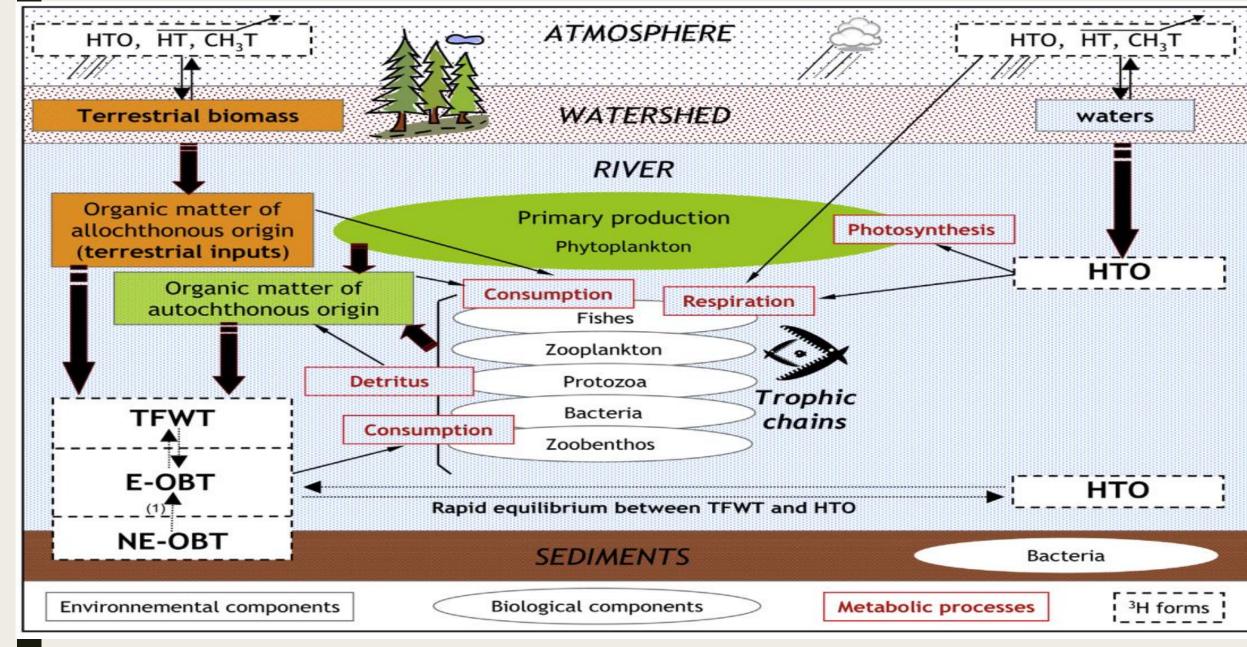
#### WHAT IS TRITIUM?

- Tritium (<sup>3</sup>H) is heavy isotope of hydrogren. It is pure beta emitter with energy 18.3 keV and it has half-life 12.3 years.
- Tritium production :
  - from nature, it called cosmogenic tritium.
  - Nuclear Power Plant and Chemical Industry.



#### Free and bound forms of tritium Mobile tritium Atmosphere or hydrosphere нто Water vapor or water Sequestered tritium **Biosphere** NE-OBT E-OBT **TFWT** - Covalent bounds - Hydrogen bounds - Free water molecules to oxygen, nitrogen, sulphur (labile hydrogen) including: SR-OBT QR-OBT Mobile tritium Quasi-instantaneous equilibrium with the hydrogen of Persistence within organic compounds depending on surrounding water molecules biodegradation and recycling kinetics

Eyrolle, F., Ducros, L., Dizès, S.L., Beaugelin-seiller, K., Charmasson, S., Boyer, P., Cossonnet, C., 2018a. An updated review on tritium in the environment. J. Environ. Radioact. 181, 128–137. https://doi.org/10.1016/j.jenvrad.2017.11.001



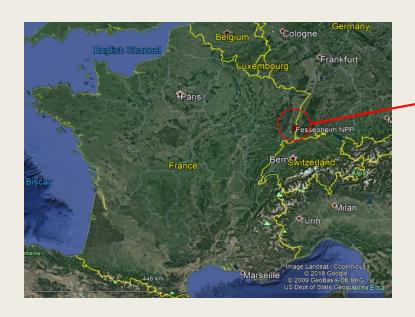
Cossonnet, C., 2014. Apparent enrichment of organically bound tritium in rivers explained by the heritage of our past 136, 162–168. https://doi.org/10.1016/j.jenvrad.2014.05.019

#### WHY WE STUDY TRITIUM?

- 1. Tritium behaviour in aquatic environment still not well known.
- 2. OBT depends on interaction with organic matter and possibly nanoparticles in river.
- 3. OBT can be accumulated in living organism, which is can make **ecotoxicity**.
- 4. To know characterization factor of tritium, we need study what parameters that can **contribute to fate factor**.
- 5. There are **no study has been done** related to tritium analysis in the Rhine River System.

# MATERIALS AND METHODS

#### **STUDY AREA**





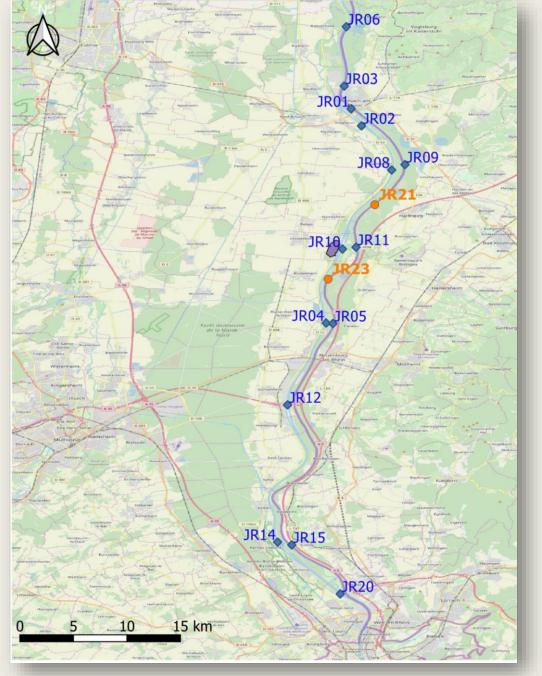
#### FESSENHEIM NPP

- PWR Type Reactor
- Has 2 Reactors that produced 900
   Mwe for each reactor
- Has been operated since 1 January 1978
- Used GCA for cooling system
- Releases 0.71 GBq/year tritium in gaseous form and 11.6 GBq/year tritium in liquid form (EDF, 2017; IAEA, 2007)
- Will be decommisioning in a few years



#### STUDY AREA

- 3 sampling points in Rhine River
- 6 sampling points in Old Rhine,
- 6 sampling points in Grand Canal.
- The samples were taken during two different months: the first campaign was in March, and the second campaign was in May



#### Sample Processing: Water Chemistry

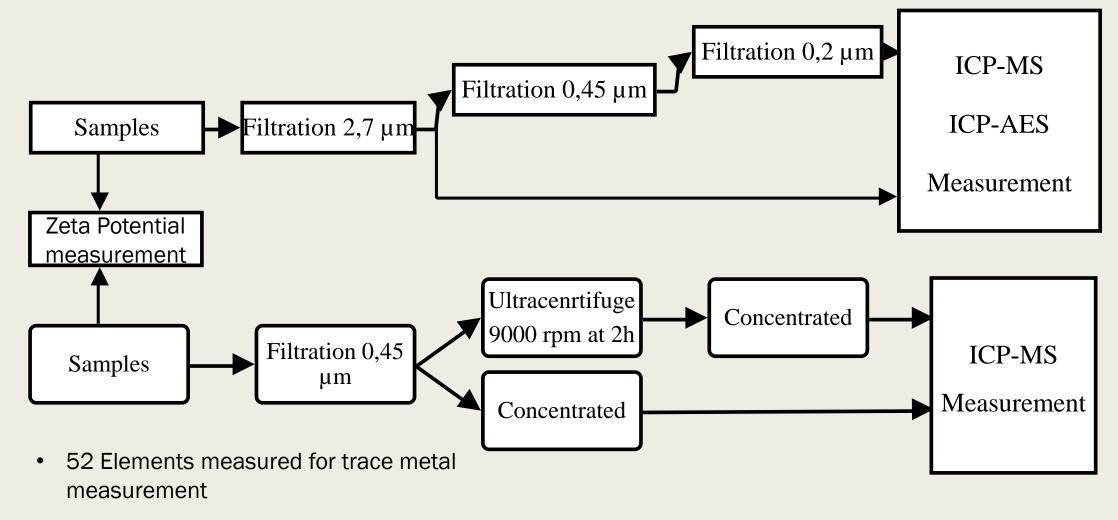
#### **On-site Measurement**

- pH
- Conductivity
- Dissolved Oxygen
- Temperature

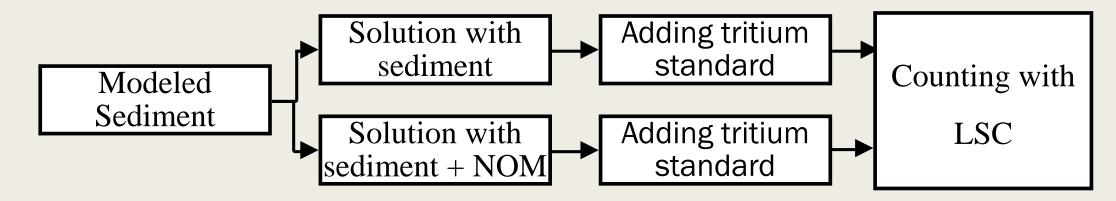
#### **In-lab Measurement**

- Total organic carbon
- Zeta Potential
- Major Elements
- Trace Metal Elements
- Anions
- Tritium

#### Sample Processing: Nanoparticles



#### Sample Processing : Sediment Experiment

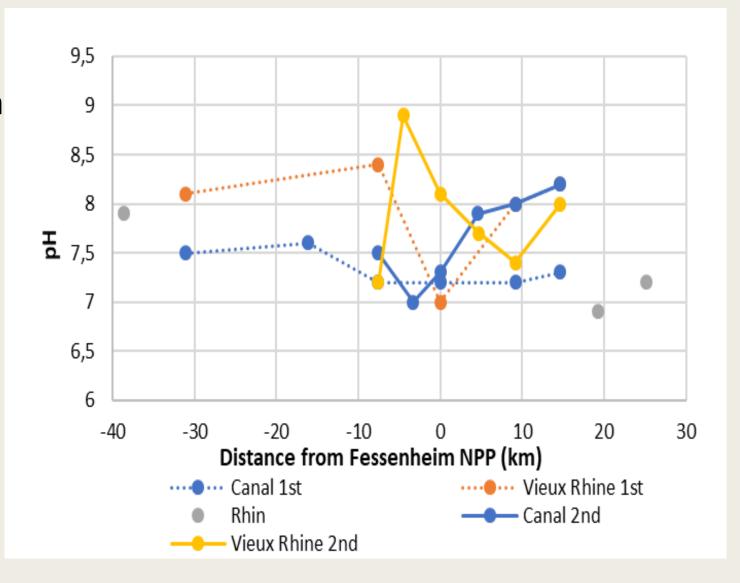


- The organic matter was collected in old Rhine (extracted and reconcentrated).
- Tritium standar  $\rightarrow$  10 µL , Activities = 74.1 Bq/L
- Ratio of sediment and organin matter = 0.2 mgC/L

### WATER CHEMISTRY

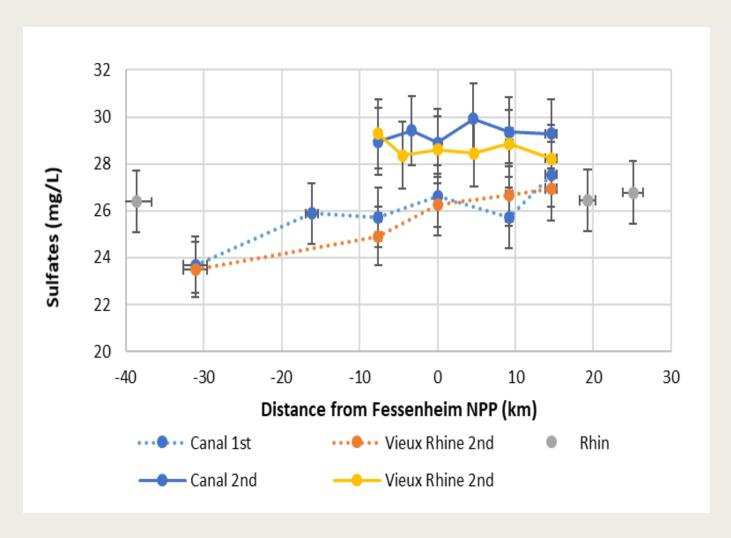
#### ph MEASUREMENT

- pH varies in the both system
- There is a possibility of a connection or exchange between the GCA and Rhine river
- pH range on both systems
   are 7 8,5
- Temperature does not vary



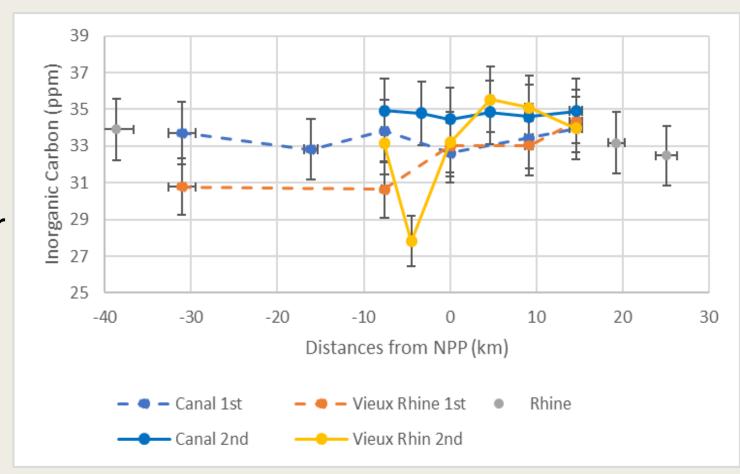
#### ANIONS MEASUREMENT

- Major Anions: sulfates and chlorine
- The evolution of major anions in both system are the same
- Anions is more concentrate in Old Rhine



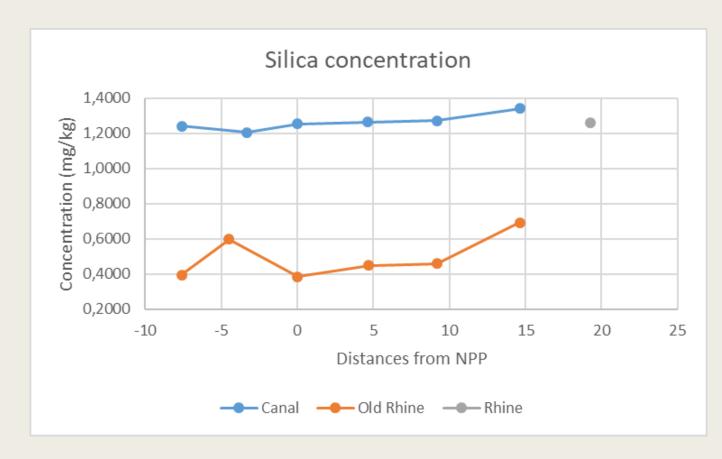
#### INORGANIC CARBON MEASUREMENT

- The concentration is the same from upstream to downstream.
- Both system have similar inorganic concentration
- Both system are highly carbonated water.



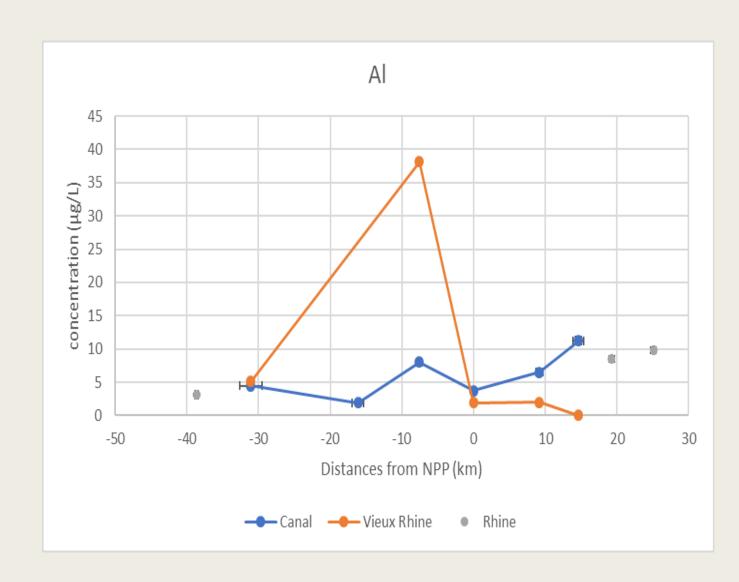
#### MAJOR ELEMENT CONCENTRATION

- Both systems have the similar evolution of major elements concentration (Al and Si).
- Si concentration is higher in the Grand Canal, as for the aluminum, the concentration is not different.



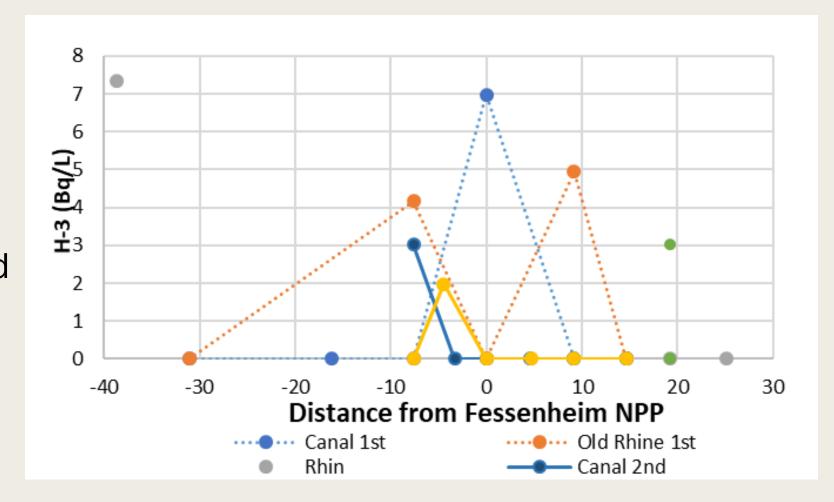
#### MAJOR ELEMENT CONCENTRATION

- Aluminum concentration is higher in old Rhine before the NPP, and the concentration decreases significantly.
- After the NPP, aluminum concentration in the old Rhine is lower than the aluminum concentration in GCA.



#### TRITIUM MEASUREMENT

- Tritium
   concentration is
   dynamic in both
   systems.
- Tritium can be found near the Fessenheim NPP.
- Other sources of tritium.



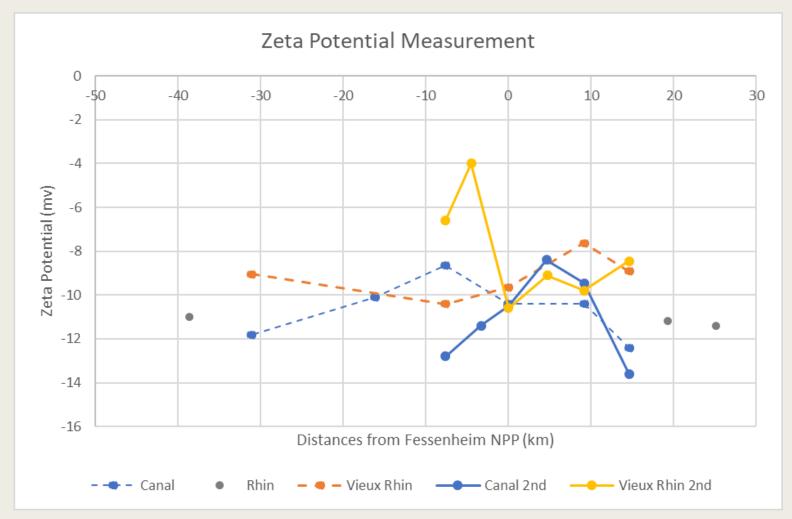
## WHAT WE LEARNED FROM WATER CHEMISTRY

- pH shows that there is a possibility that GCA and old Rhine communicate or have an exchange near the NPP.
- Both system are carbonated water.
- The same evolution of:
  - Anions
  - Trace Metal
  - Major Elements
- The tritium concentration is varies along the old Rhine and GCA

### NANOPARTICLES

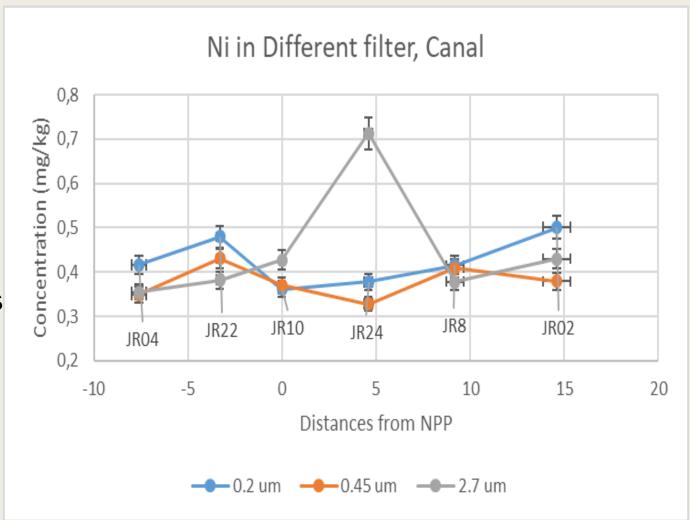
#### ZETA POTENTIAL

- Both systems have negative zeta potential values from -8 mV to -14 mV.
- The particle may not unstable and tend to make aggregates.

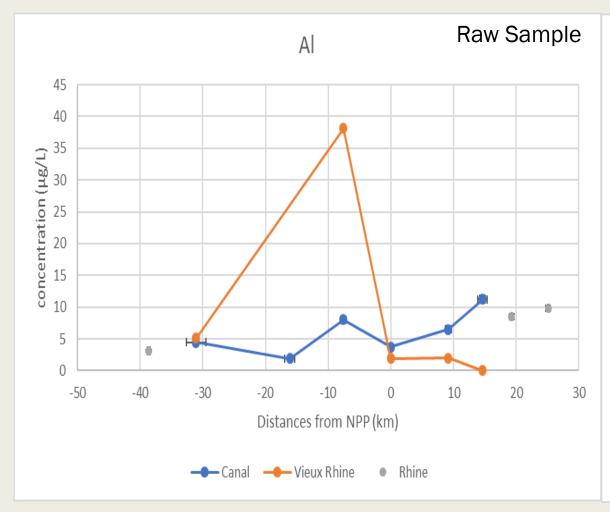


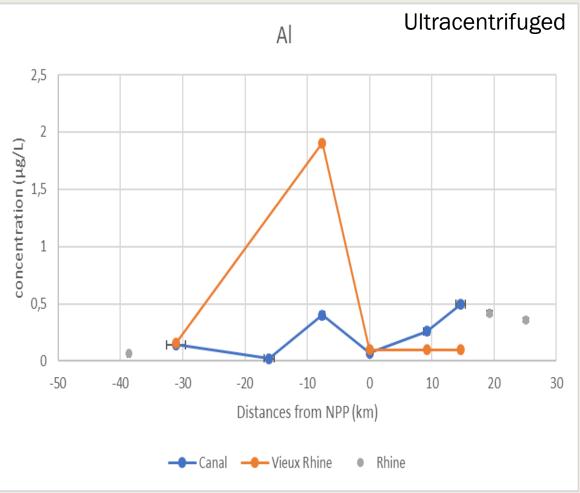
#### FILTRATION SAMPLE

- Possibility of nanoparticle present in near Fessenheim NPP.
- The same behavior found in data from the copper and zinc concentration.
- possibility to have nanoparticles (Ni,Cu and Zn) in GCA near Fessenheim NPP.



#### ULTRACENTRIFUGE SAMPLE





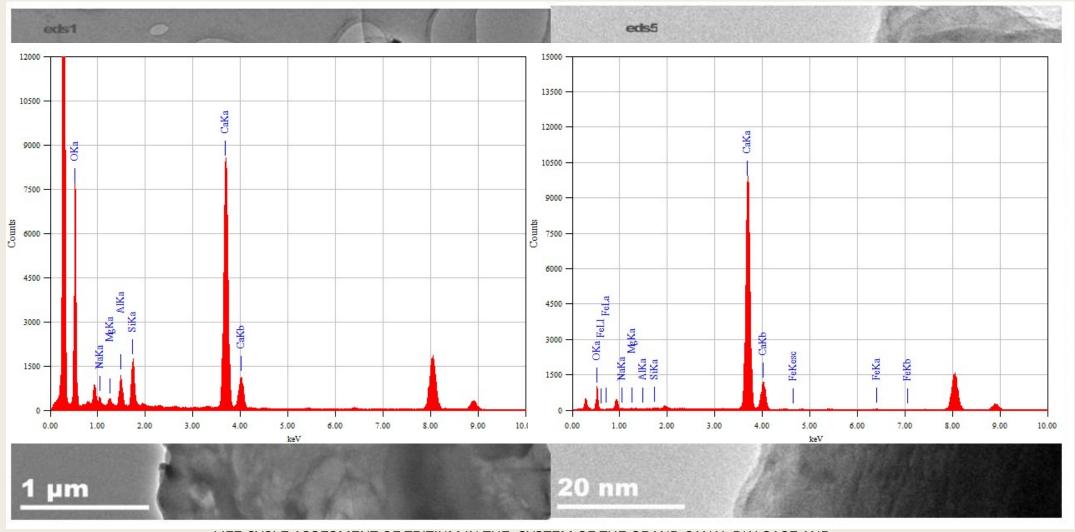
Same Trend after centrifuged → indication of nanoparticle present

#### TRITIUM DISTILLATED SAMPLES

Sample Name	Location	Raw Sample Activities (Bq/L)	Distillated Sample Activities (Bq/L)	
JR4	Canal	3.0 ± 1.3	2.2 ± 1.2	
JR23	Old Rhine	2.0 ± 1.3	<ld< th=""></ld<>	

- Grand Canal Alsace → free Tritium
- Old Rhine → bounded tritium

## RHINE RIVER RAW SAMPLES (UPSTREAM NPP)

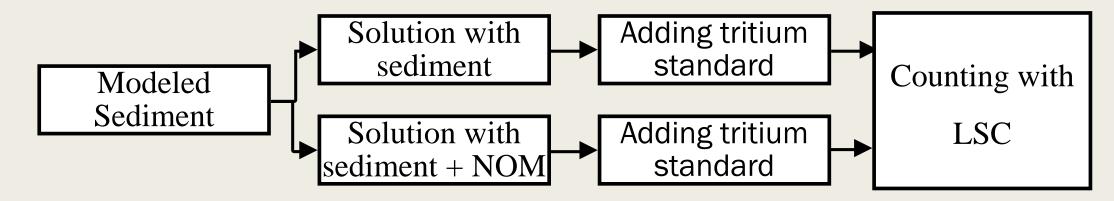


## WHAT WE LEARNED ABOUT NANOPARTICLES

- GCA and Old Rhine → zeta potential: low negative surface charge
- Filtration sample → nanoparticle of zinc, nickel and copper in GCA, 5 km after NPP.
- Ultracentrifuge sample → nanoparticle of aluminum in old Rhine, 10 km before NPP.
- TEM analysis -> nanoparticle are present in old Rhine.

## SEDIMENT ADSORPTION EXPERIMENT

#### SEDIMENT ADSORPTION EXPERIMENT



- The organic matter was collected in old Rhine (extracted and reconcentrated).
- Tritium standard  $\rightarrow$  10 µL, Activity = 74.1 Bq/L
- Ratio of sediment and organic matter = 0.2 mgC/m<sup>2</sup>

#### SEDIMENT ADSORPTION EXPERIMENT

Solution	Activity of tritium added (Bq)	NOM added (mg)	mass of sediment added (mg)	Adsorption (%)	рН
Sediment	0,669	-	21	80	6.7
Sediment +NOM	0,717	0,19	20	86	7.2

→ Organic matter increases adsorption.

# CONCLUSION AND RECOMMENDATION

#### CONCLUSION

- Water chemistry → GCA and old Rhine system have same evolution (pH, IC, Anions, Major Elements) BUT have different form of tritium.
- Nanoparticles (smectite and calcite) are present in the system → old Rhine: bounded tritium can be associated with these nanoparticles.
- Sediment adsorption experiment → tritium can be absorbed by sediment from the old Rhine river BUT the organic matter increase the tritium adsorption.

#### RECOMMENDATION

- LCA: These Results are not sufficient to estimate tritium fate factors.
- Possibly other factor :
  - pH.
  - Ratio of organic matter and sediment.
  - Composition of organic matter.
  - others possibly factor

#### SKILL USED

- Improved lab skill
- Improved presentation skill with wider audiences
- Intrepetate and analysis data
- Able to communicate and discus well with the team
- Learn new knowledge

## THANK YOU